

Bulletproof Materials

For centuries, people have used tough clothing materials to protect themselves. These materials ranged from layers of fibers to heavy metal plates and tough new materials such as synthetic fiber composites and cermets.

The use of quilted and padded material for protection replaced simple hides at about the same time woven fabrics replaced treated skins for clothes. The ancient Egyptians created light and practical bodily protection by superimposing layers of prepared and woven flax. As early as 600 A.D., the Chinese developed armor of padded silk layers. Europeans copied the technique of quilted armor from the Saracens, who brought it from the East.

About the 14th century, mounted warriors began to wear heavy armor made of steel plates. While this protected nearly every part of a knight's body, the armor was also so heavy that the man could barely move, and so costly that only the wealthy could afford a suit of armor.

Infantrymen needed much lighter protection because they required a high degree of mobility. Footsoldiers, archers, musketeers, and pikemen typically wore quilted jackets reinforced with leather or pieces of metal or bone. In the 15th century, Louis XI passed an ordinance requiring all his footsoldiers to wear linen coats that were 30 layers thick, with an option to cover the top with deerskin.

When conquistadors such as Pizarro and Cortez reached the Americas in the 15th century, they encountered Indians who wore bodily protection made of quilted cotton jackets studded with overlapping hardwood plates. This proved so effective against the conquistadors' weapons that they adopted the armor themselves.

With the increasing use of gunpowder in warfare, armored garments had to protect the wearers from bullets. In Elizabethan times, a protective doublet—called a "jack"—was made from plates of iron sewn between folds of linen; it weighted 21½ pounds. Padded, layered armor was also introduced. As described by one English lawyer: "An abundance of silken backplates and breastplates were made and sold that were pretended to be pistol proof in which any man dressed was safe as in a house, for it was impossible that any one could strike at him for laughing, so ridiculous was the figure."

However, in certain circumstances the

bulletproof jackets were indeed effective. One Swiss jacket on display in New York's Metropolitan Museum consists of nine layers of linen, weighing only three and one-half pounds; within the layers is imbedded a lead bullet that did not penetrate completely, but merely forced the inner surface to bulge. Presumably, the wearer survived what would otherwise have been a fatal wound.

The First World War used a large amount of high-explosive artillery shells, which led to a high percentage of wounds from flying shell fragments. Some special-purpose troops were issued experimental torso armor made of steel and fiber layers designed to be impervious to bullets, but this armor proved too heavy for general use.

During World War II, 80% of all casualties were caused by shell fragments, and 70% of all wounds affected the torso. Great efforts were made by both the military and industry to develop lightweight, efficient body armor. Ground troops and bomber crews wore experimental vests made of resin-bonded fiberglass, steel, aluminum, and heavy nylon cloth.

Protective vests for aircraft personnel were called "Flak" suits, from the German for "anti-aircraft artillery." These became popular at the end of World War II and were heavily used in the Korean War. Flak vests contained plates of manganese steel or a laminate of glass fabric bonded with a polyester resin, trade named Doron. In the Korean War alone, the use of body armor is credited with reducing casualties by 20%.

Semiflexible vests made of plates and basket-weave nylon were used after 1951 by the U.S. army and navy. Such vests protected against mortar fragments and artillery fragments, but could not stop armor-piercing bullets. After 1967, the use of titanium plates in body armor gave further protection.

A typical modern nylon bulletproof vest is composed of 16-24 layers of heavy-weave nylon fabric, quilt-stitched together. Ordinary bullets from submachine guns or pistols strike the outermost layers and deform into a mushroom shape that cannot penetrate the remaining layers, which dissipate its kinetic energy. Sixteen-layer vests can stop normal submachine gun or pistol bullets; 24-layer vests can stop magnum bullets. The wearer is generally bruised after such an impact, but receives no serious injury.

During the Vietnam war, pilots and crew of hovering helicopters required better protection against heavy ground fire. For this purpose, dual-hardness steel and ceramic/plastic composites were developed in the 1960s.

Dual-hardness steel gave ballistic protection 50% better than an equal weight of aluminum armor, as well as being able to absorb multiple hits better than shatterable ceramic armor. A tough crack-resisting rear face is metallurgically bonded to a very hard front face; the front face shatters the steel core of an incoming armor-piercing bullet while the ductile rear face absorbs the kinetic energy. The plates can be rolled to the thickness required to protect against a specific threat.

Ceramic composite armor is much lighter, consisting of hard ceramic plates reinforced with woven glass roving bonded with an elastic adhesive (or other plastic backing). The hardness of the ceramic slows the bullet abruptly by dissipating the projectile's energy as it shatters the ceramic. Ceramic tiles in a bulletproof vest must be replaced, though, because they shatter with every impact.

Aluminum oxide is the cheapest ceramic suitable for armor; boron carbide is the hardest such material, but is also the most expensive; silicon carbide provides a middle alternative, with hardness similar to boron carbide but less expensive. Tiles made from these ceramics are hot-pressed in graphite molds at temperatures higher than 2000°C and pressures greater than 2,000 psi.

Recent ceramic-metal composite materials, called "cermets," are five times more crack-resistant and shatter-resistant than conventional ceramics. The formation of cermets involves infiltrating molten-reactive metals into chemically treated boron carbide, boron, or boride powders or fibers. The resulting cermets have an unprecedented combination of hardness, toughness, and lightness. The properties of cermets are still being studied, and constitutive models of the material responses are under development. Researchers continue to refine the cermet-formation process to develop less expensive, commercially feasible production methods so cermets will attract more commercial exploitation. In the near future, cermet body armor could be the most effective form of bulletproof vest on the market.

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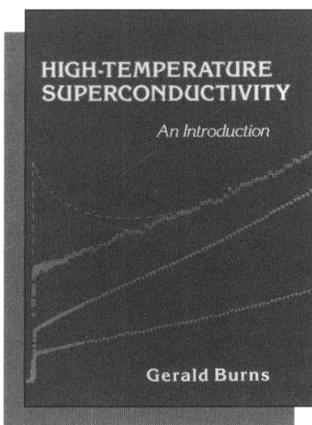
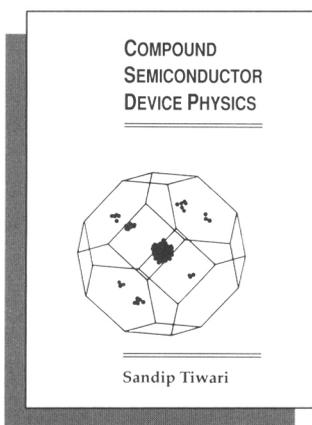
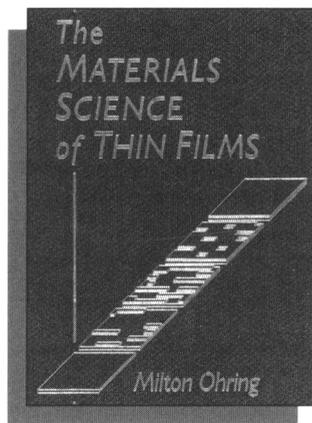
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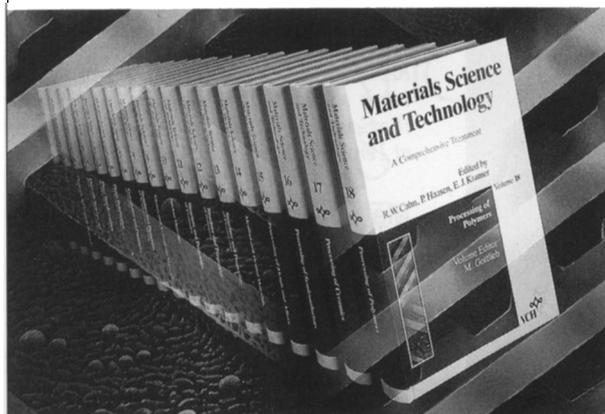
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BOOK REVIEWS

Fatigue of Materials

S. Suresh

(Cambridge University Press, 1991,
520 pages).

ISBN: 0-521-36510-4

This comprehensive, up-to-date textbook includes a substantial treatment of fatigue in ceramics and polymers as well as in metallic materials. The emphasis is on the basic science of fatigue, although the short but useful final chapter includes design considerations and case studies. A basic knowledge of dislocations is assumed, but there is a brief introduction to continuum mechanics, a more substantial introduction to linear elastic fracture mechanics, and a helpful initial overview. Topics on which much attention has been focused in recent years are well covered, such as crack closure, threshold conditions, and the short crack problem. The extensive bibliography indicates that this is a very comprehensive and up-to-date treatment, especially from the fundamental and mechanistic points of view of the subject. Indeed, if used in a course on fatigue, some degree of selection and possibly reordering of the material would probably be required. However, this well-written book certainly will be a major reference in a field

that continues to be of major importance in engineering and materials science.

Reviewer: G.W. Groves is a lecturer in the Department of Materials at the University of Oxford.

Sol-Gel Science: The Physics and Chemistry of Sol-Gel Processing

C. Jeffrey Brinker and George W. Scherer

(Academic Press, 1990, 881 pages).

ISBN: 0-12-134970-5

Over the past 15 years, sol-gel processing and synthesis of materials has grown from oddity status into a major interdisciplinary research and technological activity encompassing chemistry, physics, chemical engineering, and materials science. The rapid emergence of the field has resulted in the dispersion of its archival literature in a variety of conference proceedings and journals.

In this book, the authors have admirably accomplished their objective of presenting "a coherent account of the principles of sol-gel processing." As the first textbook-quality publication describing the entire field of sol-gel science, the authors have

had the opportunity (or onerous task) of presenting the fundamental principles underlying the unique characteristics of sol-gel systems.

The book is organized in the order of sol-gel processing, with the early chapters describing the chemical synthesis principles underlying hydrolysis and condensation. Subsequent chapters progress through gelation, aging of gels, deformation and flow, drying, structural evolution, surface chemistry, chemical modification, and sintering. The last three chapters discuss more specialized topics, such as comparisons of gel-derived and conventionally produced ceramics, film formation, and applications. The phenomenology and fundamentals of each topic are thoroughly discussed and extensively referenced with more than 100 references following every chapter.

Each chapter is an excellent account of the specific topic, not surprising since the authors are pioneers in the sol-gel field and have contributed greatly to the development of its scientific foundations. A noteworthy feature is the critical analysis of topics and the authors' attempt to point out areas of controversy as well as those requiring more research.

At \$140, the book is somewhat expensive