Material Concepts in Surface Reactivity and Catalysis

Henry Wise and Jacques Oudar (Academic Press, 1990, 260 pages) ISBN: 0-12-759940-1

With their book, Material Concepts in Surface Reactivity and Catalysis, Wise and Oudar have produced a unique contribution to the legion of publications that dwell on surface science and catalysis. They begin by addressing the salient properties of bonding and structure in bulk crystals and at crystalline surfaces, with emphasis on how surface structure affects the deployment of adspecies. The formalisms that describe crystal surface energy and shape are presented and discussed in the context of understanding metal/support interactions and preferred crystal orientation. Fundamental concepts for interpreting and applying classical adsorption isoterms in the study of surface adsorption processes are elaborated upon, followed by a phenomenological treatment of adsorption and desorption kinetics that includes a description of traditional measurement methods. The authors also examine the principles and procedures commonly employed to investigate adsorbate binding energy and structure-low-energy electron diffraction and x-ray and ultraviolet photoemission spectroscopy-and tie this in to the intriguing topic of adsorbate-induced surface reconstruction.

Subsequent chapters address reactions at metal/support interfaces, electronic properties of nonmetal catalysts, and disorder in multicomponent metal oxides, followed by a discussion of the properties of metal oxides that influence specific activity and selectivity during catalysis. Much of the emphasis in this set of chapters is on binary oxides having the spinel, perovskite, and scheelite structures.

The final three chapters cover (1) grain boundary properties at surfaces, including grain boundary energetics, segregation effects, and adsorption at grain boundaries; (2) oxide layer formation at metal/gas interfaces; and (3) adsorption at metal/electrolyte interfaces. Prominent attention is paid to the kinetic aspects of oxide layer

growth in gaseous environments and to the potential dependence of adsorption processes in electrolytic media.

The authors do an excellent job of weaving fundamental concepts and corresponding descriptive expressions together with illustrations from experimental measurements. Where thermodynamic principles are involved, they are presented in appreciable detail. The book is replete with informative illustrations (tables and figures) that drive home the main points of discussion. Unfortunately, some of the figures are blurred or so poorly reproduced that they are only partly legible.

One shortcoming is that the references are not more current than the early 1980s; in fact, a majority of the references are pre-1980. In a rapidly evolving field like surface science, some readers will tend to wonder how up to date the contents actually are. While some of the latest thinking from the decade of the 1980s on the relationship between materials properties and surface reactivity is surely missing from Wise and Oudar's tome, much of the information they present is based on principles and concepts that are accepted, time-tested axioms.

On the whole, Material Concepts in Surface Reactivity and Catalysis would be a useful addition to the personal library of any scientist or engineer with a strong interest in surface science, heterogeneous catalysis, and the solid state in general. It could also serve very nicely as a course book for an advanced undergraduate and graduate level curriculum, particularly for interdepartmental/multidisciplinary courses. Indeed, physists, chemists, and materials scientists alike would gain valuable insights about surface science and the solid state from a course of study based on Wise and Oudar's book.

Reviewer: Victor A. Maroni is a senior chemist working jointly in the Materials Science Division and the Chemical Technology Division of Argonne National Laboratory. His research interests include molecular sieve catalysis, aqueous corrosion, and the application of spectroscopic methods to the study of materials properties.

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Positions Available

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Applicants should forward a resume, including names and addresses of three or more references to: Prof. Neil M. Hawkins, Head of Civil Engineering, University of Illinois, 1114a Newmark CE Laboratory, 205 N. Mathews Avenue, Urbana, Illinois 61801-2397; telephone: (217) 333-3815; FAX (217) 333-9464. This full-time position is available January 1, 1992; priority will be given to applications received by September 30, 1991. The salary range is \$32,000 to \$36,000.

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Positions Available

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The Ames Laboratory invites applications and nominations for the position of Deputy Director. The Ames Laboratory is a national laboratory of the Department of Energy (DOE), dedicated to research in fundamental and applied science and engineering related to the nation's current and future energy concerns, and is operated for DOE by Iowa State University (ISU). The Deputy Director has primary responsibility for new initiatives, the Office of Information, the Office of Internal Auditing, the scientific computing group, and all scientific program reviews. The Deputy Director will often represent the Laboratory at local, state and national forums and will act as Laboratory Director in the Director's absence.

The successful candidate will hold a PhD in physical science or engineering, have a scientific record commensurate with faculty status in a related academic department of ISU, have significant experience in science administration, and have a strong working knowledge of the Department of Energy.

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Research Associate needed by solar energy research company in Golden, CO to perform molecular-beam massspectrometric (MBMS) studies of biomass, municipal solid waste and its components. Upgrade fast pyrolysis oil through zeolite catalyst with MBMS. Perform catalytic destruction on biomass gasifier tars using MBMS. Requires PhD, physical organic chemistry; proven ability in characterization of organic compounds using triple quarupole mass spectrometry, including specific studies in organic transients via collisioninduced dissociation (CID) and ionmolecule reaction. Ability may be gained through employment or educational program. \$36,000/year; 8:00am - 5:00pm, M-F. Respond by resume to Colorado Department of Labor & Employment, Division of Employment & Training, 600 Grant, Suite 900, Denver, CO 80203, ATT: James Shimada, and refer to Job Order No. C03195434.

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Department of Nuclear Engineering

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Please send CV and references to: Mr. Steve Carlson Personnel Services University of California Santa Barbara, CA 93106-3160

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RESEARCH ASSOCIATE

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Contact: Prof. Gregory L. Griffin
Department of Chemical
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